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States ...~~

8 Chemistry: Matter and Change • Chapter 5 Teaching
Transparency Masters z x y 2s orbital z x y 1s orbital z x px py pz
p orbitals d orbitals y z x y z x y z x y dxy dxz dyz dx²-y² dz² x y
y x z x y z z z x y Atomic Orbitals TEACHING TRANSPARENCY
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Transparency Activity Masters A teacher support and planning
section including Content Outline of the chapter Spanish
Resources Answers and teacher notes for the worksheets Hands-

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On Activities MiniLAB and Lab Worksheets: Each of these worksheets is an expanded version of each lab and MiniLAB found in the Student Edition.

~~UNIT 1 RESOURCES Earth Science~~

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 9 11 . Soluble or Insoluble? Use your solubility table to determine if the following chemicals are soluble or insoluble in water. Write the terms "aqueous" or "solid" next to each chemical. 1>.k507.) 1.

~~Chapters 1-5 Resources~~

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 9 11 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g.

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silver, lead d. zinc, sodium 2.

~~LESSON PLAN 12—Glencoe/McGraw Hill, Inc.~~

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Name _____ Date _____ Class _____
Formation of Ions 1. What are the names of the two elements shown? Date Class 22 Use with Chapter 7, Section 7.1
2. Are the elements shown on the left sides of the two equations neutral?

~~Name _____ Date _____ Class _____ TEACHING TRANSPARENCY~~

TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2
2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ulevel? 4.

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Section Focus Transparency Masters Math Skills Transparency

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Masters and Worksheets Teaching Transparency Masters and Worksheets Additional Resources: Solving Problems: A Chemistry Handbook Cooperative Learning in the Science Classroom Lab and Safety Skills in the Science Classroom glencoe.com Lab Resources: Laboratory Manual OL CBL Laboratory ...

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an excellent review instrument. Answers to the Study Guide pages are included in the Teacher Guide and Answers section at the back of this book. Reinforcement: These pages provide

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opportunities that complete your teaching cycle and benefit all your students. Reinforcement masters are especially helpful for students who require additional

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~~Chapter 1: Introduction to Chemistry~~

60 Chemistry: Matter and Change • Chapter 3 Teaching Transparency Masters Heterogeneous mixtures dirt, blood, milk

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lemonade, gasoline, steel oxygen, gold, iron salt, baking soda, sugar Homogeneous mixtures Compounds Elements Matter Mixtures Can it be separated by physical means? Is the composition uniform? Can it be broken down by ordinary ...

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36 Chemistry: Matter and Change • Chapter 14 Teaching
Transparency Masters 0 10 20 30 40 50 60 70 80 90 100
Solubility (g solute/100 g H₂O) 90 100 80 70 60 50 40 30 20 10
0 Temperature (oC) Solubilities as a Function of Temperature
NaCl KClO₃ KCl CaCl₂ Ce₂(SO₄)₃ Solubility-Temperature
Graphs TEACHING TRANSPARENCY MASTER Use with ...

~~TEACHING TRANSPARENCY MASTER 42 Solubility Temperature ...~~
TEACHING TRANSPARENCY WORKSHEET Ionic Bonds Use with
Chapter 7, Section 7.2 1. How many valence electrons does a
neutral magnesium (Mg) atom have? 2. What is the charge on a

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magnesium ion? What does magnesium have to do to form such an ion, and why does it tend to do so? 3. How many valence electrons does a single neutral chlorine (Cl atom) have? 4.

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TEACHING TRANSPARENCY 22 Chemistry: Matter and Change

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TEACHING TRANSPARENCY 22 Chemistry: Matter and Change

Teaching Transparency Worksheet 2 Use with Chapter 7,

Formation of Ions Section 7.1 1. ... TEACHER GUIDE AND

ANSWERS Chemistry: Matter and Change 3 Teacher Guide and
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~~TEACHING TRANSPARENCY MASTER 11 Cathode Ray~~

~~Experiments ...~~

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~~Transparency Masters Chemistry: Matter and Change • Chapter 11 31 number of moles of known substance C H C H C H O O O CO CO CO H O H O H O number of moles of unknown substance O CO H O C H CO H O C H O H O C H O CO use mole ratio 5 mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol C 3 H 8 4 mol H 2 O 1 mol ...~~

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18-2 Connecting Math to Physics FAST FILE Chapters 16-20
Resources, Chapter 18 ... Analysis Answers will vary. Sample
answers: The

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