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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite soluble in water (a) HCl c an ionic compound that is not very soluble in water (b) NaNO₃

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CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl₂ does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl₂ is an ionic solid. In the crystal form, its ions are locked in position.

13 Ions in Aqueous Solutions and Colligative Properties

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. The element boron, B, has an atomic mass of 10.81 amu according to the periodic table.

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Chapter 12 Mixed Review Solutions Modern Chemistry

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CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

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