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elementary electrochemistry in somewhat greater depth than is found in standard textbooks, but at a level still suitable for first-year college and advanced high school courses. It should also be helpful as review material for students in more advanced courses in analytical and physical chemistry.

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Chapter 17 Holt chemistry vocab. STUDY. PLAY. Oxidation, a reaction that removes one or more electrons from a substance such that the substance's valence or oxidation state increase. ... an electrochemical device in which electrolysis takes place when an electric current is in the device.

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~~Holt Chemistry Chapter 17: Oxidation, Reduction, and ...~~

CHAPTER SEVENTEEN ELECTROCHEMISTRY For Review 1. Electrochemistry is the study of the interchange of chemical and electrical energy. A redox (oxidation-reduction) reaction is a reaction in which one or more electrons are transferred.

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6. Answer the following questions about electrochemistry. (a) Several different electrochemical cells can be constructed using the materials shown below. Write the balanced net-ionic equation for the reaction that occurs in the cell that would have the greatest positive value of E_{cell} .
 $\text{Al(s)} \rightarrow \text{Al}^{3+}(\text{aq}) + 3 \text{e}^{-}$
 $\text{Cu}^{2+}(\text{aq}) + 2 \text{e}^{-} \rightarrow \text{Cu(s)}$

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~~CHAPTER SEVENTEEN ELECTROCHEMISTRY—Cengage~~

Electrochemistry Balancing in Basic Solution • If a reaction occurs in basic solution, one can balance it as if it occurred in acid. • Once the equation is balanced, add OH^{-} to each side to “neutralize” the H^{+} in the equation and create water in its place. • If this produces water on both sides, you might have to subtract water ...

~~Unit 7 Electrochemistry—Nelson~~

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